

## Chemistry in Everyday Life

\* Chemotherapy → The branch of a chemistry which deals with treatment of disease using suitable chemical substances is called chemotherapy.

Drug and medicine →

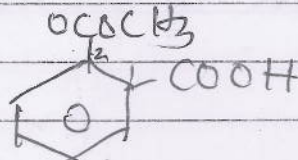
Medicine → is a chemical substance which causes useful effect to prevent and cure various diseases without side effects. They do not cause addiction.

Drug → is a chemical substance which also cures the disease but is habit forming and causes various side effects.

A substance when taken as prescribed by doctor act as medicine but when the substance is taken in large dose it will act as drug.

Antipyretics → The chemical substances which are used to lower the temp. of a body in high fever is called antipyretics.  
eg. Paracetamol, Aspirin

Aspirin →



2-Acetoxy benzoic acid



**Analgesic** → The chemical substances which are used to relieve pain without causing disturbance to nervous system is known as analgesic.  
eg → Aspirin

**Antimicrobials** → The chemical substances used to cure the diseases caused by microorganism such as bacteria are known as antimicrobials.

**Antibiotics** → These are the chemical substances which are produced by microorganism or chemical synthesis and can inhibit the growth or even kill the microorganism.

**Antifertility** → These are the chemical substances used to control the pregnancy. They are also known as contraceptives and birth control drugs.

**Antacid** → These are the chemical substances which neutralize the excess of acid in our body and give relief from acidity and heart burn.  
eg. Baking soda in water is very common antacid.

**Antiseptic and disinfectant** →

**Antiseptic** → The chemical substances which are used either to kill or prevent the growth of microorganism are known as antiseptics.



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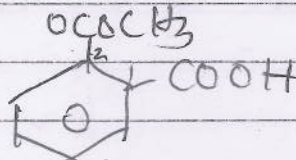
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Medicine → is a chemical substance which causes useful effect to prevent and cure various diseases without side effects. They do not cause addiction.

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Dettol mix. of chloroxylenol & terpineol

They can be applied to the living tissue:

Disinfectant → The chemical substance which are used to kill the microorganisms are called as disinfectant.

Disinfectant cannot be used on living tissues.

A same substance can act as an antiseptic and disinfectant.

eg → 0.2% phenol solution is antiseptic  
1% phenol solution is disinfectant.

Antihistamine → These are the chemical substances which diminish or abolish the main action of histamine released in the body and hence prevent the allergic rxn.

\* Chemicals in food → There are certain chemicals which are added to food to improve its keeping qualities, taste, and appearance - these are called as food adhesives.

Preservatives → These are the chemical substances which are used to protect the food from microorganisms such as bacteria, yeast etc and retain their nutritional value.

eg → Na benzoate

Artificial sweetening agent → The chemical



Substances which gives sweetening effect to the food and et change its flavour are called as artificial sweetening agent.

Eg → Saccharin, Aspartame, Sucralose, Alitame

**Antioxidants** → These are the chemical substances used to prevent the oxidation of fats in processed food such as potato chips, biscuits etc.

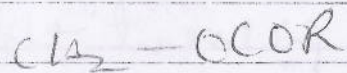
Eg → Butylated hydroxy toluene -

\* **Soap and detergent** →

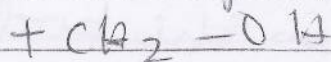
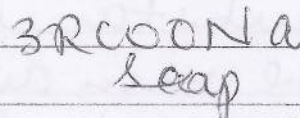
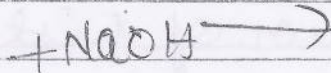
Chemically soap is Na salt of fatty acid ( $\text{RCOONa}$ )

Soap can be prepared by the process of saponification.

In this process oil or fats (esters of fatty acids) are heated with  $\text{NaOH}$



Tri fatty acid



**Disadvantages of soap** →

① It can't be used in hard water

② It can't be used in acidic solution

**Advantages of soap** →

① It is biodegradable.

**Detergent** → Chemically detergent are the Na salts of long chain Sulphonates



or sulphates. Generally represented by  
 $R-SO_3-NA$

Advantages of detergent  $\rightarrow$

- ① It can be used in hard water.
- ② It can be used in acidic solution.
- ③ Its cleaning action is more effective than soap.

Disadvantages of detergent  $\rightarrow$

- ① It is non biodegradable and causes environment pollution.

Types of detergent  $\Rightarrow$

- ① Cationic detergent  $\rightarrow$  contains cationic hydrophilic group
- ② Anionic detergent
- ② Non ionic detergent